=> fil req

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=> d que stat 18 L2 STR

VAR G1=7/SI/N/P/O/S

NODE ATTRIBUTES:

NSPEC IS RC AT 1
NSPEC IS RC AT 5
NSPEC IS RC AT 6
NSPEC IS RC AT 7
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 7

.....

STEREO ATTRIBUTES: NONE
L4 93768 SEA FILE=REGISTRY SSS FUL L2
L5 STR

November 13, 2009 10/540,558

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VAR G1=7/SI/N/P/O/S
NODE ATTRIBUTES:
NSPEC IS RC AT 5
NSPEC IS RC AT 6
NSPEC IS RC AT 7
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED
GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 7
STEREO ATTRIBUTES: NONE
               SCR 2043 OR 2049
L8
            222 SEA FILE=REGISTRY SUB=L4 SSS FUL L5 NOT L6
100.0% PROCESSED 294 ITERATIONS
                                                             222 ANSWERS
SEARCH TIME: 00.00.01
=> d que stat 111
L2
               STR
                   C @ 7
VAR G1=7/SI/N/P/O/S
NODE ATTRIBUTES:
NSPEC IS RC AT 1
NSPEC IS RC
                AT 5
NSPEC IS RC AT 6
NSPEC IS RC AT 7
DEFAULT MLEVEL IS ATOM
DEFAULT ECLEVEL IS LIMITED
GRAPH ATTRIBUTES:
RING(S) ARE ISOLATED OR EMBEDDED
NUMBER OF NODES IS 7
STEREO ATTRIBUTES: NONE
L4 93768 SEA FILE=REGISTRY SSS FUL L2
                SCR 2043 OR 2049
L6
1.9
                STR
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VAR G1=8/12/16/19/23
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VAR G2=O/S/7/SI/N/P NODE ATTRIBUTES:

NSPEC IS RC AT 5 NSPEC IS RC AT 6 NSPEC IS RC AT 7 AT 10 NSPEC IS RC NSPEC IS RC AT 11 AT 13 NSPEC IS RC IS RC AT 17 NSPEC IS RC NSPEC AT 20 NSPEC IS RC AT 21 NSPEC IS RC AT 22 DEFAULT MLEVEL IS ATOM

DEFAULT ECLEVEL IS AIOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 28

STEREO ATTRIBUTES: NONE

L11 186 SEA FILE=REGISTRY SUB=L4 SSS FUL L9 NOT L6

100.0% PROCESSED 265 ITERATIONS SEARCH TIME: 00.00.01 186 ANSWERS

3

DEFINION TIME: CO.CO.C.

=> d his

(FILE 'HOME' ENTERED AT 13:45:26 ON 13 NOV 2009)

FILE 'LREGISTRY' ENTERED AT 13:58:25 ON 13 NOV 2009

L1 STR L2 STR L1

FILE 'REGISTRY' ENTERED AT 14:12:19 ON 13 NOV 2009

L3 50 S L2 L4 93768 S L2 FUL

FILE 'LREGISTRY' ENTERED AT 14:40:14 ON 13 NOV 2009 L5 STR L2

FILE 'REGISTRY' ENTERED AT 15:28:19 ON 13 NOV 2009

4

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1.6
              SCR 2043 OR 2049
            13 S L5 NOT L6 SSS SAM SUB=L4
L8
           222 S L5 NOT L6 SSS FUL SUB=L4
               SAV L8 WEI558S1/A
    FILE 'LREGISTRY' ENTERED AT 15:30:05 ON 13 NOV 2009
L9
               STR L5
    FILE 'REGISTRY' ENTERED AT 15:37:03 ON 13 NOV 2009
           13 S L9 NOT L6 SSS SAM SUB=L4
L11
           186 S L9 NOT L6 SSS FUL SUB=L4
               SAV L11 WEI558S2/A
L12
            36 S L8 NOT L11
    FILE 'HCAPLUS' ENTERED AT 15:38:39 ON 13 NOV 2009
L13
            32 S L12
T.14
           58 S L11
L15
            1 S 2004:570217/AN
L16
           84 S L13 OR L14
             QUE ELECTROLY?
L17
L18
              OUE BATTERY
           10 S L16 AND L17
L19
L20
            6 S L16 AND L18
L21
           10 S L19 OR L20
L22
              OUE ADDITIVE? OR ADJUVANT? OR AUXILIAR? OR MODIF?
L23
             QUE AGENT?
L24
            2 S L16 AND L22
L25
            1 S L16 AND L23
L26
           11 S L21 OR L24-25
=> fil hcap
FILE 'HCAPLUS' ENTERED AT 15:48:46 ON 13 NOV 2009
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FILE 'HCAPLUS' ENTERED AT 15:48:46 ON 13 NOV 2009
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FILE COVERS 1907 - 13 Nov 2009 VOL 151 ISS 21
FILE LAST UPDATED: 12 Nov 2009 (20091112/ED)
REVISED CLASS FIELDS (/NCL) LAST RELOADED: Aug 2009
USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Aug 2009
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HCAplus now includes complete International Patent Classification (IPC) reclassification data for the third quarter of 2009.

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This file contains CAS Registry Numbers for easy and accurate substance identification.

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=> d ibib abs hitstr hitind 126 1-22

L26 ANSWER 1 OF 11 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2007:1334246 HCAPLUS Full-text

DOCUMENT NUMBER: 147:544588

TITLE: Nonaqueous electrolyte containing

phosphazene compound and lithium ion secondary

battery with high discharge efficiency

having the same

INVENTOR(S): Nakagawa, Hiroe; Katayama, Sadahiro; Nukuta,

Toshivuki

PATENT ASSIGNEE(S): GS Yuasa Corp., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 16pp.

CODEN: JKXXAF DOCUMENT TYPE: Patent

LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2007305551	A	20071122	JP 2006-135814	200605
PRIORITY APPLN. INFO.:			JP 2006-135814	200605 15
INIONIII ALPEN. INFO.:			01 2000 103014	200605 15

OTHER SOURCE(S): MARPAT 147:544588

AB Disclosed is a nonaq. electrolyte made from an organic material consisting of a lithium salt, and a salt at molten state at room temperature containing a (cyclic) phosphazene compound and a quaternary ammonium organic cation.

850650-07-6

RL: TEM (Technical or engineered material use); USES (Uses) (nonag. electrolyte containing phosphazene compound for lithium ion secondary battery with high discharge

efficiency) 850650-07-6 HCAPLUS

Phosphoramidofluoridic acid,

N-[difluoro[(trifluorophosphoranylidene)amino]phosphoranylidene]-, ethvl ester (CA INDEX NAME)

$$\texttt{EtO} = \bigoplus_{i=1}^{O} \texttt{N} = \bigoplus_{i=1}^{F} \texttt{N} = \texttt{PF3}$$

nonaq electrolyte cyclic phosphazene compd lithium ion secondary battery; quaternary ammonium org cation

Secondary batteries

(lithium; nonaq. electrolyte containing phosphazene compound for lithium ion secondary battery with high discharge efficiency)

Battery electrolytes

(nonag. electrolyte containing phosphazene compound for lithium ion secondary battery with high discharge

Quaternary ammonium compounds, uses IT

> RL: TEM (Technical or engineered material use); USES (Uses) (nonag. electrolyte containing phosphazene compound for lithium ion secondary battery with high discharge efficiency)

33027-66-6 90076-65-6, LiTFSi 132843-44-8, Lithium bis(perfluoroethanesulfonvl)imide 143314-16-3, 1-Ethyl-3-methylimidazolium tetrafluoroborate 174501-64-5, 1-Butv1-3-methylimidazolium hexafluorophosphate 174501-65-6, 1-n-Butyl-3-methylimidazolium tetrafluoroborate

850650-07-6 RL: TEM (Technical or engineered material use); USES (Uses) (nonaq. electrolyte containing phosphazene compound for lithium ion secondary battery with high discharge efficiency)

L26 ANSWER 2 OF 11 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2007:145522 HCAPLUS Full-text DOCUMENT NUMBER: 146:232676

TITLE: Lithium secondary batteries suppressing electrolytes from

decomposing at high temperature and their cathodes and cathode materials

INVENTOR(S): Ichihashi, Akira; Kano, Gentaro; Okawa, Takeshi PATENT ASSIGNEE(S): Sony Corp., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 17pp.

CODEN: JKXXAF DOCUMENT TYPE: Patent

LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 2007035391 A 20070208 JP 2005-215427 200507 PRIORITY APPLN. INFO.: JP 2005-215427 200507 26

OTHER SOURCE(S): MARPAT 146:232676

The battery cathodes contain materials coated with phosphazenes preferably represented by $X3(X1X2P:N)nX4[X1-X4 = F, C1, Br, alkoxy, phenyl(oxy); n <math>\geq 4$]. The cathodes may contain Li- and transition metal-containing active mass compds. Batteries containing the cathodes, anodes, and electrolytes inside film-type packages are also claimed. The batteries show less swelling on hightemperature uses.

924658-15-1 924658-17-3 924658-23-1

November 13, 2009 10/540,558

924658-25-3 924658-27-5 924658-28-6

RL: TEM (Technical or engineered material use); USES (Uses) (active-mass coatings; battery cathodes having phosphazene coatings on active masses and suppressing decomposition at

phosphazene coatings on active masses and suppressing decomposition at high temperature)

RN 924658-15-1 HCAPLUS

CN Phosphorodifluoridimidic acid,

 $\label{eq:normalized} \mbox{N-}(\mbox{P},\mbox{P-}\mbox{difluoro-N-}\mbox{methoxyphosphinimy1})-, methy1 ester (CA INDEX NAME)$

RN 924658-17-3 HCAPLUS

CN Phosphorodifluoridimidic acid, N-[N-[N-[N-(P,P-difluoro-N-methoxyphosphinimyl)-P,P-difluorophosphinimyl]-P,P-difluorophosphinimyl]-P,P-difluorophosphinimyl]-, methyl ester (CA INDEX NAME)

RN 924658-23-1 HCAPLUS

N Phosphorodifluoridimidic acid,

RN 924658-25-3 HCAPLUS

CN Phosphorodifluoridimidic acid.

N-[N-[N-(P,P-difluoro-N-propoxyphosphinimyl)-P,P-difluorophosphinimyl]-P,P-difluorophosphinimyl]-P,P-difluorophosphinimyl]-, propyl ester (CA INDEX NAME)

- RN 924658-27-5 HCAPLUS
- CN Phosphorodifluoridimidic acid,

N-[N-[N-[N-(P,P-difluoro-N-phenoxyphosphinimyl)-P,Pdifluorophosphinimyl]-P,P-difluorophosphinimyl]-P,Pdifluorophosphinimyl]-, phenyl ester (CA INDEX NAME)

$$\mathtt{PhO} - \overbrace{\overbrace{f}}^{\overline{f}} = \mathtt{N} - \overbrace{\overbrace{f}}^{\overline{f}} = \mathtt{N} - \overbrace{\overbrace{f}}^{\overline{f}} = \mathtt{N} - \mathtt{OPh}$$

924658-28-6 HCAPLUS

CN Phosphoramidimidic difluoride,

N'-(P,P-difluoro-N-phenylphosphinimyl)-N-

[[[(difluorophenylphosphoranylidene)aminoldifluorophosphoranylidene]amino]difluorophosphoranylidene]- (CA INDEX NAME)

$$\text{Ph-} \underbrace{\bar{f}}_{} = \text{N-} \underbrace{\bar{f}}_{} = \text{N-} \underbrace{\bar{f}}_{} = \text{N-} \underbrace{\bar{f}}_{} = \text{N-} \text{Ph}$$

- 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
- lithium secondary battery cathode phosphazene coating; polydifluorophosphazene battery cathode coating high temp swelling prevention
- Battery electrolytes

(battery cathodes having phosphagene coatings on active masses and suppressing decomposition at high temperature)

- Polyphosphazenes
 - RL: TEM (Technical or engineered material use); USES (Uses) (chlorine-containing, active-mass coatings; battery

cathodes having phosphazene coatings on active masses and

suppressing decomposition at high temperature)

- Polyphosphazenes
 - RL: TEM (Technical or engineered material use); USES (Uses) (fluorine-containing, active-mass coatings; battery

cathodes having phosphazene coatings on active masses and suppressing decomposition at high temperature)

Fluoropolymers, uses

RL: TEM (Technical or engineered material use); USES (Uses)

(gels, electrolytes; battery cathodes having

phosphazene coatings on active masses and suppressing decomposition at high temperature)

- Secondary batteries

(lithium; battery cathodes having phosphazene coatings on active masses and suppressing decomposition at high temperature)

- Fluoropolymers, uses
 - RL: TEM (Technical or engineered material use); USES (Uses)

(polyphosphazene-, active-mass coatings; battery

cathodes having phosphazene coatings on active masses and suppressing decomposition at high temperature)

TT 924658-15-1 924658-17-3 924658-19-5 924658-21-9 924658-23-1 924658-25-3 924658-27-5 924658-28-6

> RL: TEM (Technical or engineered material use); USES (Uses) (active-mass coatings; battery cathodes having

phosphazene coatings on active masses and suppressing decomposition at high temperature)

7782-42-5, Graphite, uses

RL: TEM (Technical or engineered material use); USES (Uses) (anodes; battery cathodes having phosphazene coatings

on active masses and suppressing decomposition at high temperature) 12190-79-3, Lithium cobaltate (LiCoO2)

RL: TEM (Technical or engineered material use); USES (Uses)

(cathode active mass; battery cathodes having phosphazene coatings on active masses and suppressing decomposition at high temperature)

ΤТ 21324-40-3, Lithium hexafluorophosphate

RL: TEM (Technical or engineered material use); USES (Uses)

(electrolytes, infiltrated in polymer gels;

battery cathodes having phosphazene coatings on active masses and suppressing decomposition at high temperature)

9011-17-0, Hexafluoropropylene-vinylidene fluoride copolymer RL: TEM (Technical or engineered material use); USES (Uses)

(gels, electrolytes; battery cathodes having

phosphazene coatings on active masses and suppressing decomposition at high temperature)

L26 ANSWER 3 OF 11 HCAPLUS COPYRIGHT 2009 ACS on STN 2005:450196 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 142:492196

TITLE: Electrolytic double-layer capacitors employing nonaqueous electrolytic

solutions and showing good charge performance

Kanno, Hiroshi; Otsuki, Masami INVENTOR(S):

PATENT ASSIGNEE(S): Bridgestone Corp., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 20 pp. CODEN: JKXXAF

DOCUMENT TYPE: Pat.ent. LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2005135951	A	20050526	JP 2003-367066	200310
PRIORITY APPLN. INFO	D.:		JP 2003-367066	28
				200310 28

- AB The capacitors, having porous carbon as electrode active masses, contain nonaq. electrolytic solns. and satisfy the ratio of leakage current before and after 60° heat stability test ≤60%. The electrolytic solns. may contain aprotic solvents and P compds. and/or N compds. The capacitors may satisfy charge voltage ≥2.7 V.
- 22474-63-1D, fluorinated, alkoxy-substituted 852178-24-6 852178-25-7

RL: DEV (Device component use); MOA (Modifier or additive use); USES

10

(Uses)

(double-layer capacitors containing oligocyclophosphazenes and showing good capacitance holding ratio)

RN 22474-63-1 HCAPLUS

CN Phosphorimidic trifluoride, (difluorophosphinyl) - (8CI, 9CI) (CA INDEX NAME)

RN 852178-24-6 HCAPLUS

CN Methanesulfonamide, N-(trifluorophosphoranylidene)- (CA INDEX NAME)

RN 852178-25-7 HCAPLUS

CN Acetic acid, 2-[fluoro[(trifluorophosphoranylidene)amino]phosphinyl]-, methyl ester (CA INDEX NAME)

IT 852178-23-5

RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)

(electrolytic solns.; double-layer capacitors containing oligocyclophosphazenes and showing good capacitance holding ratio)

RN 852178-23-5 HCAPLUS

CN Phosphoramidic difluoride, (difluoro-1-piperidinylphosphoranylidene)(9CI) (CA INDEX NAME)

$$\text{response} = \text{response}$$

IC ICM H01G009-038

CC 76-10 (Electric Phenomena)

ST electrolytic double layer capacitor leakage current

minimized; charge performance electrolytic double layer capacitor; static capacitance holding electrolytic capacitor

IT Electrolytic capacitors

(double-layer; double-layer capacitors containing oligocyclophosphazenes and showing good capacitance holding ratio)

IT 22474-63-1D, fluorinated, alkoxy-substituted

852178-24-6 852178-25-7

RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)

(double-layer capacitors containing oligocyclophosphazenes and showing good capacitance holding ratio)

IT 852178-23-5

RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)

(electrolytic solns.; double-layer capacitors containing oligocyclophosphazenes and showing good capacitance holding ratio)

L26 ANSWER 4 OF 11 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2005:445414 HCAPLUS Full-text

ACCESSION NUMBER: 2005:44541 DOCUMENT NUMBER: 142:492192

TITLE: Electrolytic double-layer capacitors

employing nonagueous electrolytic

solutions

INVENTOR(S): Kanno, Hiroshi; Otsuki, Masatomo

PATENT ASSIGNEE(S): Bridgestone Corp., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 19 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2005135950	A	20050526	JP 2003-367052	200310
PRIORITY APPLN. INFO.:			JP 2003-367052	28 200310 28

AB The capacitors contain nonaq. electrolytic solns. preferably containing P compds. and/or N compds. and satisfy static capacitance degradation $\leq 10\%$ on heat stability test at 60%. The electrolytic solns. may contain aprotic organic solvents. The pos. and neg. electrodes of the capacitors may contain porous carbon (of surface functional group number ≤ 100 meg/g) as active masses. The capacitors show charge voltage of ≥ 2.5 V and long-term stability of capacitance performance.

I 22474-63-1D, ethoxy-substituted derivs. 852178-23-5 852178-24-6 852178-25-7

RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)

(electrolytic solns.; electrolytic

double-layer capacitors containing cyclic oligophosphazenes in nonaq. electrolytic solns.)

RN 22474-63-1 HCAPLUS

N Phosphorimidic trifluoride, (difluorophosphinyl) - (8CI, 9CI) (CA INDEX NAME) 12

RN 852178-23-5 HCAPLUS

CN Phosphoramidic difluoride, (difluoro-1-piperidinylphosphoranylidene)-(9CI) (CA INDEX NAME)

RN 852178-24-6 HCAPLUS

CN Methanesulfonamide, N-(trifluorophosphoranylidene)- (CA INDEX NAME)

RN 852178-25-7 HCAPLUS

CN Acetic acid, 2-[fluoro[(trifluorophosphoranylidene)amino]phosphinyl]-, methyl ester (CA INDEX NAME)

IC ICM H01G009-038

ICS H01G009-058

CC 76-10 (Electric Phenomena)

ST electrolytic double layer capacitor charge performance; nonaq electrolytic soln phosphazene contg capacitor; porous carbon electrode capacitor long term stability

IT Electrolytic capacitors

(double-layer; electrolytic double-layer capacitors containing cyclic oligophosphazenes in nonaq. electrolytic solns.)

IT Carbon black, uses
RL: DEV (Device component use); USES (Uses)

(electrolytic double-layer capacitors containing cyclic oligophosphazenes in nonaq. electrolytic solns.)

Polyphosphazenes

RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)

(oligomeric, cyclic, fluorinated, alkoxy-substituted;

electrolytic double-layer capacitors containing cyclic

oligophosphazenes in nonag. electrolytic solns.)

7440-44-0, Carbon, uses

RL: DEV (Device component use); USES (Uses)

(electrodes; electrolytic double-layer capacitors

containing cyclic oligophosphazenes in nonag, electrolytic solns.)

108-32-7, Propylene carbonate

RL: DEV (Device component use); USES (Uses)

(electrolytic double-layer capacitors containing cyclic

oligophosphazenes in nonag. electrolytic solns.)

22474-63-1D, ethoxy-substituted derivs. 852178-23-5 852178-24-6 852178-25-7

RL: DEV (Device component use); MOA (Modifier or additive use); USES

(Uses)

(electrolytic solns.; electrolytic

double-layer capacitors containing cyclic oligophosphazenes in nonaq.

electrolytic solns.)

OS.CITING REF COUNT: THERE ARE 1 CAPLUS RECORDS THAT CITE THIS 1 RECORD (1 CITINGS)

L26 ANSWER 5 OF 11 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2005:368511 HCAPLUS Full-text DOCUMENT NUMBER: 142:433056

TITLE:

Secondary nonaqueous electrolyte

batterv INVENTOR(S): Koto, Tomoko

PATENT ASSIGNEE(S):

Japan Storage Battery Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 13 pp.

KIND DATE

CODEN: JKXXAF DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION: PATENT NO

	FAIRNI NO.	KTMD	DAIL	AFFEICATION NO.	DAIL
	JP 2005116306	A	20050428	JP 2003-348133	
					200310
					07
PRIO	RITY APPLN. INFO.:			JP 2003-348133	
					200310

AB The battery has a cathode, containing a Li-Ni-Mn composite oxide : Lix NiyMn2 $v04-\delta$ (0< x< 1.1; 0.45< vr< 0.55; and 0< $\delta<$ 0.4) as a cathode active mass, an anode, and a nonag, electrolyte solution; where the electrolyte solution contains 0.1-20 mass%, phosphazene derivative

APPLICATION NO

DATE

0.7

850650-07-6

RL: MOA (Modifier or additive use); USES (Uses)

(cathodes containing lithium manganese nickel oxides and electrolytes containing phosphazene derivs. for secondary lithium batteries)

RN 850650-07-6 HCAPLUS

CN Phosphoramidofluoridic acid,

N-[difluoro[(trifluorophosphoranylidene)amino]phosphoranylidene]-, ethyl ester (CA INDEX NAME)

14

IC ICM H01M010-40

ICS H01M004-02; H01M004-58

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

ST secondary lithium battery cathode lithium manganese nickel

oxide; battery electrolyte phosphazene deriv

IT Battery cathodes

Battery electrolytes

(cathodes containing lithium manganese nickel oxides and electrolytes containing phosphazene derivs. for secondary

lithium batteries)

IT Polyphosphazenes

RL: MOA (Modifier or additive use); USES (Uses)

(cyclic; cathodes containing lithium manganese nickel oxides and electrolytes containing phosphazene derivs. for secondary

lithium batteries)

IT Secondary batteries

(lithium; cathodes containing lithium manganese nickel oxides and electrolytes containing phosphazene derivs. for secondary lithium batteries)

T 96-48-0, γ-Butyrolactone 96-49-1, Ethylene carbonate

105-58-8, Diethyl carbonate 108-32-7, Propylene carbonate 616-38-6, Dimethyl carbonate 623-53-0, Ethyl methyl carbonate

7782-42-5, Graphite, uses 12031-75-3, Lithium manganese nickel oxide (LiMn1.5Ni0.504) 14283-07-9, Lithium tetrafluoroborate

21324-40-3, Lithium hexafluorophosphate

RL: DEV (Device component use); USES (Uses)

(cathodes containing lithium manganese nickel oxides and electrolytes containing phosphazene derivs. for secondary

lithium batteries)

IT 850650-07-6

RL: MOA (Modifier or additive use); USES (Uses)

(cathodes containing lithium manganese nickel oxides and electrolytes containing phosphazene derivs. for secondary lithium batteries)

OS.CITING REF COUNT: 3

THERE ARE 3 CAPLUS RECORDS THAT CITE THIS

RECORD (3 CITINGS)

L26 ANSWER 6 OF 11 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2004:605979 HCAPLUS Full-text

DOCUMENT NUMBER: 141:149554

TITLE: Separator for nonaqueous-electrolyte

double layer capacitor

INVENTOR(S): Kanno, Hiroshi; Otsuki, Masami; Eguchi, Shinichi

PATENT ASSIGNEE(S): Bridgestone Corp., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 18 pp.

CODEN: JKXXAF

10/540.558 15

DOCUMENT TYPE: LANGUAGE:

Patent Japanese FAMILY ACC, NUM, COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2004214356	A	20040729	JP 2002-381018	
				200212 27
PRIORITY APPLN. INFO.:			JP 2002-381018	
				200212 27

OTHER SOURCE(S): MARPAT 141:149554

- A nonflammable separator for a nonaq.-electrolyte double layer capacitor comprises a finely porous film formed by adding a phosphazene derivative (or its isomer) to a polymer. Specifically, the polymer may comprise a polyolefin such as polyethylene or polypropylene.
- 722454-84-4 722454-85-5 722454-86-6
 - RL: DEV (Device component use); USES (Uses)

(separator containing phosphazene derivative for nonaq.-

electrolyte double layer capacitor)

722454-84-4 HCAPLUS RN

CN Phosphoramidic acid, (trifluorophosphoranylidene)-, diethyl ester (9CI) (CA INDEX NAME)

- 722454-85-5 HCAPLUS RN
- CN Phosphonimidic difluoride, N-(methylsulfonyl)-P-1-pyrrolidinyl-(9CI) (CA INDEX NAME)

$$\begin{array}{c} F \\ F \\ \downarrow \\ N \\ \downarrow \\ N \\ \end{array}$$

- RN 722454-86-6 HCAPLUS
- CN Phosphorodifluoridimidic acid, acetyl-, phenyl ester (9CI) (CA INDEX NAME)

IC ICM H01G009-02

CC 76-10 (Electric Phenomena)

ST phosphazene deriv polymer separator nonaq electrolyte double layer capacitor

IT Capacitors

(double layer; separator containing phosphazene derivative for nonaq.electrolyte double layer capacitor)

IT Porous materials

(films; separator containing phosphazene derivative for nonaq.-electrolyte double laver capacitor)

IT Films

(porous; separator containing phosphazene derivative for nonaq.electrolyte double layer capacitor)

IT Polvolefins

RL: DEV (Device component use); USES (Uses)

(separator containing phosphazene derivative for nonaq.electrolyte double layer capacitor)

IT Phosphazenes

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(separator containing phosphazene derivative for nonaq.-

electrolyte double layer capacitor)
II 2397-48-0 9002-88-4, Polyethylene 9003-07-0, Polypropylene

72454-84-4 72454-85-5 722454-86-6 724792-60-3

RL: DEV (Device component use); USES (Uses)

(separator containing phosphazene derivative for nonaq.electrolyte double layer capacitor)

L26 ANSWER 7 OF 11 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2004:589783 HCAPLUS Full-text

DOCUMENT NUMBER:

PATENT INFORMATION:

141:126373

TITLE: Separator for nonaqueous electrolyte battery

INVENTOR(S): Kanno, Hiroshi; Otsuki, Masashi; Equchi,

Shinichi

PATENT ASSIGNEE(S): Bridgestone Corporation, Japan SOURCE: PCT Int. Appl., 32 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1

PAT	TENT	NO.			KIN	D	DATE			APPL	ICAT	ION :	NO.		D	ATE
						-										
		-														
WO	2004	0620	02		A1		2004	0722		WO 2	003-	JP16	360			
															2	00312
															1	9
	W:	ΑE,	AG,	AL,	AM,	AT,	AU,	AZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,
		CH,	CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,
		GB,	GD,	GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,
		KR,	KZ,	LC,	LK,	LR,	LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,

MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM,

	DK, SE,	BY, EE, SI, NE,	ES, SK,	FI, TR,	FR, BF,	GB,	GR,	HU,	IE,	IT,	LU,	MC,	NL,	PT,	RO,
AU	20032894	53		A1		2004	0729	ž	AU 2	003-	2894	53			00312
EP	1603175			A1		2005	1207	1	EP 2	003-	7809	36			00312
	R: AT, PT, SK	BE, IE,													
CN	1732580			A		2006	0208	(CN 2	003-	8010	7738			00312
US	20060073	381		A1		2006	0406	τ	US 2	005-	5408	37			00506
	7585587 APPLN.	INFO.	:	В2		2009	0908	ć	JP 2	002-	3806	83	i	A. 2	00212
								ī	wio 2	003-	JP16	360	1		00312

- AB The separator, which is incombustible even when the inside of a battery has a high temperature and useful for a primary or secondary Li battery, comprises a micro-porous film formed by adding a phosphazene derivative and/or an isomer of a phosphazene derivative to a polymer.
- IT 722454-84-4 722454-86-6 724792-59-0

RL: DEV (Device component use); USES (Uses)

(separators containing phosphazene derivative added polymers for primary and secondary lithium batteries)

- RN 722454-84-4 HCAPLUS
- CN Phosphoramidic acid, (trifluorophosphoranylidene)-, diethyl ester (9CI) (CA INDEX NAME)

- RN 722454-86-6 HCAPLUS
- CN Phosphorodifluoridimidic acid, acetyl-, phenyl ester (9CI) (CA INDEX NAME)

- RN 724792-59-0 HCAPLUS
- CN Phosphonimidic difluoride, N-(methylsulfonyl)-P-1-piperidinyl- (9CI) (CA INDEX NAME)

$$\underbrace{ \sum_{i=1}^{N} N - \bigcup_{i=1}^{N} - Me}_{N-i}$$

- ICM H01M002-16
 - ICS H01M010-40
- CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
- ST nonag electrolyte battery incombustible
- separator phosphazene deriv added polymer Primary battery separators
- Secondary battery separators

(separators containing phosphazene derivative added polymers for primary and secondary lithium batteries)

- ΤТ 7439-93-2, Lithium, uses
 - RL: DEV (Device component use); USES (Uses)

(anode; separators containing phosphazene derivative added polymers for primary and secondary lithium batteries)

- IT 1313-13-9, Manganese dioxide, uses 12190-79-3, Cobalt lithium oxide (CoLiO2)
 - RL: DEV (Device component use); USES (Uses)

(cathode; separators containing phosphazene derivative added polymers for primary and secondary lithium batteries)

96-48-0, y-Butyrolactone 96-49-1, Ethylene carbonate

105-58-8, Diethyl carbonate 957-13-1 1184-10-7 2397-48-0 9002-88-4, Polyethylene 14283-07-9, Lithium tetrafluoroborate

33027-68-8 722454-84-4 722454-86-6

724792-59-0

RL: DEV (Device component use); USES (Uses)

(separators containing phosphazene derivative added polymers for primary and secondary lithium batteries)

ΤТ 724792-60-3

RL: DEV (Device component use); USES (Uses) 2

(separators containing phosphagene derivative added polymers for primary and secondary nonag. electrolyte batteries)

REFERENCE COUNT:

THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L26 ANSWER 8 OF 11 HCAPLUS COPYRIGHT 2009 ACS on STN 2004:570217 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 141:126304

TITLE: Additive for secondary battery nonaqueous electrolyte solution and

the battery

INVENTOR(S): Otsuki, Masashi; Horikawa, Yasuro

PATENT ASSIGNEE(S): Bridgestone Corporation, Japan

SOURCE: PCT Int. Appl., 33 pp.

CODEN: PIXXD2 DOCUMENT TYPE: Patent

LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

						KIND DATE								Е	ATE	
	2004		82		A1		2004	0715		WO 2	003-	JP16	592			00312
	W:	CH, GB, KR, MX, SG,	CN, GD, KZ, MZ, SK,	CO, GE, LC, NI,	CR, GH, LK, NO, SY,	CU, GM, LR, NZ, TJ,	AU, CZ, HR, LS, OM, TM,	DE, HU, LT, PG,	DK, ID, LU, PH,	DM, IL, LV, PL,	DZ, IN, MA, PT,	EC, IS, MD, RO,	EE, JP, MG, RU,	EG, KE, MK, SC,	BZ, ES, KG, MN, SD,	CA, FI, KP, MW, SE,
	RW:	BW, AZ, DK, SE,	GH, BY, EE, SI,	GM, KG, ES,	KE, KZ, FI, TR,	LS, MD, FR, BF,	MW, RU, GB, BJ,	TJ, GR,	TM, HU,	AT, IE,	BE, IT,	BG, LU,	CH, MC,	CY, NL,	CZ,	DE, RO,
AU	2003						2004	0722		AU 2	003-	2927	64			00312
EP	1580	832			A1		2005	0928		EP 2	003-	7681	80		2	00312 4
EP	1580 R:	AT,	BE,	CH,	DE,	DK,	2009 ES, FI,	FR,								
CN	1732	588			A		2006	0208		CN 2	003-	8010	7739			00312
CN US	1003 2006	6268 0046	9 151												_	00506
PRIORIT	Y APP	LN.	INFO	.:						JP 2	002-	3771	42	i	A. 2	00212 6
										WO 2	003-	JP16	592	1	2	00312

AB The additive comprises a phosphazene derivative represented by R13P = N-X (R1 = halo or monovalent substituent; and X = C, Si, N, P, O and/or S containing organic group). The battery has a nonaq. electrolyte solution comprising the above additive, a cathode, and an anode.

IT 722454-84-4 722454-85-5 722454-86-6

RL: MOA (Modifier or additive use); USES (Uses)
(additives containing phosphazene derivs. for secondary
battery electrolytes)

RN 722454-84-4 HCAPLUS

CN Phosphoramidic acid, (trifluorophosphoranylidene)-, diethyl ester (9CI) (CA INDEX NAME)

RN 722454-85-5 HCAPLUS

CN Phosphonimidic difluoride, N-(methylsulfonyl)-P-1-pyrrolidinyl-(9CI) (CA INDEX NAME)

RN 722454-86-6 HCAPLUS

CN Phosphorodifluoridimidic acid, acetyl-, phenyl ester (9CI) (CA INDEX NAME)

IC ICM H01M010-40

C 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)

ST secondary lithium battery nonaq electrolyte

additive phosphazene deriv

IT Battery electrolytes

(additives containing phosphazene derivs. for secondary battery electrolytes)

IT Secondary batteries

(lithium; additives containing phosphazene derivs. for secondary battery electrolytes)

IT 96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate

12190-79-3, Cobalt lithium oxide (CoLiO2) 21324-40-3, Lithium hexafluorophosphate

RL: DEV (Device component use); USES (Uses)

(additives containing phosphazene derivs. for secondary battery electrolytes)

IT 2397-48-0 722454-84-4 722454-85-5 722454-86-6

RL: MOA (Modifier or additive use); USES (Uses)

(additives containing phosphazene derivs. for secondary

battery electrolytes)

REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L26 ANSWER 9 OF 11 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2004:570177 HCAPLUS Full-text

November 13, 2009 10/540,558 21

DOCUMENT NUMBER: TITLE:

141:132681 Phosphazene derivative additives for nonaqueous electrolytic solution and nonaqueous electrolyte electric

double-layer capacitors INVENTOR(S): PATENT ASSIGNEE(S):

Otsuki, Masashi; Horikawa, Yasuro Bridgestone Corporation, Japan PCT Int. Appl., 31 pp.

SOURCE: DOCUMENT TYPE:

Patent Japanese

CODEN: PIXXD2

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

LANGUAGE:

PATENT NO.					KIN		DATE			APPL	ICAT	ION	NO.		DATE		
	2004		71		A1		2004	0715		WO 2	003-	JP16	585			200312	
	W:	CH, GB, KR, MX, SG,	CN, GD, KZ, MZ, SK,	CO, GE, LC, NI, SL,	CR, GH, LK, NO, SY,	CU, GM, LR, NZ, TJ,	CZ, HR, LS, OM,	DE, HU, LT, PG,	DK, ID, LU, PH,	DM, IL, LV, PL,	DZ, IN, MA, PT,	EC, IS, MD, RO,	EE, JP, MG, RU,	EG, KE, MK, SC,	BZ ES KG MN SD	24 , CA, , FI, , KP, , MW, , SE, , VC,	
	RW:	BW, AZ, DK, SE,	BY, EE,	GM, KG, ES, SK,	KE, KZ, FI, TR,	LS, MD, FR, BF,	RU, GB,	TJ, GR,	TM, HU,	AT, IE,	BE, IT,	BG, LU,	CH, MC,	CY, NL,	CZ PT	, AM, , DE, , RO, , ML,	
AU	2003	2927.	58		A1		2004	0722		AU 2	003-	2927	58			200312 24	
EP	1577	913			A1		2005	0921		EP 2	003-	7681	73			200312	
	R:															, MC, , HU,	
CN	1732				A		2006	0208		CN 2	003-	8010	7740			200312	
US	2006	0092	596		A1		2006	0504		US 2	005-	5405	65			200506	
)RIT	APP:	LN.	INFO	.:						JP 2	002-	3771	28		A	200212 26	
										WO 2	003-	JP16	585			200312	

The title additives in the nonaq. electrolytic solns. for elec. double-layer AB capacitors are phosphazene derivs. R13P=N (R1 = halo, monovalent substituent; X = organic group containing C, Si, N, P, O, S). The additives have high dissoln, power for supporting salts and a low viscosity. A nonag, electrolyte elec. double-layer capacitors provided with the title electrolytic solution

24

containing the additives have excellent fast or high-rate charge-discharge characteristics.

IT 722454-84-4P 722454-85-5P

722454-86-6P

RL: PNU (Preparation, unclassified); PRP (Properties); PREP (Preparation)

(phosphazene derivative additives for nonaq. electrolytic solution and nonaq. electrolyte elec. double-layer capacitors)

RN 722454-84-4 HCAPLUS

CN Phosphoramidic acid, (trifluorophosphoranylidene)-, diethyl ester (9CI) (CA INDEX NAME)

RN 722454-85-5 HCAPLUS

CN Phosphonimidic difluoride, N-(methylsulfonyl)-P-1-pyrrolidinyl-(9CI) (CA INDEX NAME)

RN 722454-86-6 HCAPLUS

CN Phosphorodifluoridimidic acid, acetyl-, phenyl ester (9CI) (CA INDEX NAME)

IC ICM H01G009-038

CC 76-10 (Electric Phenomena)

ST phosphazene additive nonaq electrolyte fast charging double layer capacitor

IT Electric double layer

(capacitors; phosphazene derivative additives for nonaq. electrolytic solution and nonaq. electrolyte elec. double-layer capacitors)

IT Phosphazenes

RL: MOA (Modifier or additive use); PRP (Properties); USES (Uses) (derivs., additives; phosphazene derivative additives for nonaq. electrolytic solution and nonaq. electrolyte elec. double-laver capacitors)

November 13, 2009 Capacitors (double layer, nonag. electrolytic solns. containing phosphazene additives; phosphazene derivative additives for nonaq, electrolytic solution and nonaq. electrolyte elec. double-layer capacitors) Dissolution (of salts, in electrolytic solution; phosphazene derivative additives for nonag. electrolytic solution and nonaq. electrolyte elec. double-layer capacitors) Electrolytic solutions (phosphazene derivative additives for nonaq. electrolytic solution and nonag, electrolyte elec. double-layer capacitors) ΤТ Viscosity (salts, in phosphazene-containing electrolyte solution; phosphazene derivative additives for nonag. electrolytic solution and nonaq. electrolyte elec. double-layer capacitors) ΙT Solubility (saturation, of salts, in electrolytic solution; phosphazene derivative additives for nonaq. electrolytic solution and nonag. electrolyte elec. double-layer capacitors) 2397-48-0P TT RL: PNU (Preparation, unclassified); PRP (Properties); PREP (Preparation) (phosphazene additive, in nonaq. electrolytic solution; phosphazene derivative additives for nonag. electrolytic solution and nonaq. electrolyte elec. double-layer capacitors) IΤ 722454-84-4P 722454-85-5P 722454-86-6P RL: PNU (Preparation, unclassified); PRP (Properties); PREP (Preparation) (phosphazene derivative additives for nonag. electrolytic solution and nonaq. electrolyte elec. double-layer capacitors) IΤ 429-06-1, Tetraethylammonium tetrafluoroborate RL: MOA (Modifier or additive use); PRP (Properties); USES (Uses) (saturation solubility in phosphazene-containing electrolytic solution; phosphazene derivative additives for nonag. electrolytic solution and nonaq. electrolyte elec. double-layer capacitors) OS.CITING REF COUNT: THERE ARE 1 CAPLUS RECORDS THAT CITE THIS 1 RECORD (1 CITINGS) REFERENCE COUNT: THERE ARE 11 CITED REFERENCES AVAILABLE 11 FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT L26 ANSWER 10 OF 11 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2004:139816 HCAPLUS Full-text DOCUMENT NUMBER: 140:184695 TITLE: Secondary nonaqueous electrolyte

batterv INVENTOR(S): Narioka, Yoshinori; Mori, Sumio

PATENT ASSIGNEE(S): Japan Storage Battery Co., Ltd., Japan SOURCE: Jpn. Kokai Tokkvo Koho, 14 pp.

CODEN: JKXXAF DOCUMENT TYPE: Patent LANGUAGE: Japanese

24

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2004055208	A	20040219	JP 2002-208280	
				200207 17
ORITY APPLN. INFO.:			JP 2002-208280	
				200207

- AB The battery has an active mass containing anode and a Li salt dissolved nonaq. electrolyte solution; where the electrolyte solution has a halo-containing phosphazene compound and the anode has a binder comprising a non-halo material.
- IT 657348-91-9

PRI

RL: DEV (Device component use); USES (Uses)

(electrolyte solns. having halo-containing phosphazene compds. for secondary lithium batteries)

RN 657348-91-9 HCAPLUS

CN Phosphinimidic fluoride, N-ethoxy-P,P-bis(pentafluoroethyl)- (9CI) (CA INDEX NAME)

- IC ICM H01M010-40
 - ICS H01M004-02; H01M004-62
- CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)
- T secondary battery electrolyte halo contg

phosphazene compd; anode binder nonhalo compd secondary battery

IT Fluoropolymers, uses

Styrene-butadiene rubber, uses

RL: DEV (Device component use); USES (Uses)

(anode binder; anode binders containing non-halo materials for

secondary lithium batteries)

IT Battery anodes

(anode binders containing non-halo materials for secondary lithium batteries)

IT Battery electrolytes

(electrolyte solns. having halo-containing phosphazene

compds. for secondary lithium batteries)

IT Secondary batteries

(secondary lithium batteries having halo-containing

phosphazene compds. in electrolyte solns. and non-halo

materials in anodes)

7782-42-5, Graphite, uses

RL: DEV (Device component use); USES (Uses)

(anode active mass; anode binders containing non-halo materials for secondary lithium batteries)

IT 24937-79-9, Pvdf

RL: DEV (Device component use); USES (Uses)

(anode binder; anode binders containing non-halo materials for secondary lithium batteries)

T 96-49-1, Ethylene carbonate 105-58-8, Diethyl carbonate 21324-40-3, Lithium hexafluorophosphate 657348-91-9 RL: DEV (Device component use); USES (Uses)

(electrolyte solns. having halo-containing phosphazene compds. for secondary lithium batteries)

IT 9003-55-8

RL: DEV (Device component use); USES (Uses)

(styrene-butadiene rubber, anode binder; anode binders containing non-halo materials for secondary lithium batteries)

OS.CITING REF COUNT: 1 THERE ARE 1 CAPLUS RECORDS THAT CITE THIS
RECORD (1 CITINGS)

L26 ANSWER 11 OF 11 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1999:327314 HCAPLUS $\underline{\text{Full-text}}$

DOCUMENT NUMBER: 131:87959

TITLE: Chemistry of Diazaphospholephosphines. 2.

Exocyclic Phosphine-Sulfido, -Selenido, and

-Imido Derivatives of a Diazaphospholephosphine System. Crystal and Molecular Structures of Two Diazaphospholephosphine Imines:

4-(Difluoro((p-cyanotetrafluorophenyl)imino)phos phorano)-2,5-dimethyl-2H-1,2,3G2-diazaphos

phorano)-2,5-dimethyl-2H-1,2,3 σ 2-diazaphos phole and

4-(Bis(dimethylamino)(((pentamethylcyclopentadie nyl)dichlorotitanio)imino)phosphorano)-2,5-dimet

 $hyl-2H-1,2,3\sigma 2-diazaphosphole$

AUTHOR(S): Mikoluk, Michael D.; McDonald, Robert; Cavell, Ronald G.

Department of Chemistry, University of Alberta,

Alberta, AB, T6G 2G2, Can.

SOURCE: Inorganic Chemistry (1999), 38(12), 2791-2801

CODEN: INOCAJ; ISSN: 0020-1669

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

ANGUAGE: Eng

$$\mathsf{Me} = \mathsf{N} \mathsf{P} \mathsf{P} \mathsf{X}_2$$

CORPORATE SOURCE:

AB The substituted-exo-phosphine diazaphospholephosphines I (X = F, NMe2, OCH2CF3) are exclusively oxidized at this center with either chalcogens (S, Se) or azides to phosphoranodiazaphospholes. Oxidation imparts a dramatic upfield shift of the phosphorus NMR signals and an increase in the 1JPC coupling consts. within the ring. (Difluorophosphino)diazaphosphole was also oxidized with selected amines using di-Et azodicarboxylate (DAD) as the coupling agent. Bulky amines (e.g., 2,4,6-tri-tert-bulylaniline (mes*)) gave the monomeric iminophosphorane whereas less bulky amines (p-toluidine) formed mostly the cyclic diazadiphosphetidine. The crystal and mol. structure of

4-(difluoro((p-cyanotetrafluorophenyl)imino)phosphorano)-2.5- dimethyl-2H- $1,2,3\sigma_2$ -diazaphosphole was determined: triclinic, P.hivin.1 (Number 2), a = 7.2744(15) Å, b = 10.087(4) Å, c = 10.566(2) Å, α = 66.62(2)°, β = 77.60(2)°, $\gamma = 78.14(3)$ °, V = 688.8(4) Å3, Z = 2. Final indexes are R1 = 0.0368 and wR2 = 0.0968, and for all data, R1 = 0.0478, wR2 = 0.1033, and GOF = 1.067. The structure revealed two planar ring systems consisting of the diazaphosphole and the p-tetrafluorophenyl (tfbn) ring with an angle of 26.3° between the rings. The angle about the phosphine imine nitrogen (i.e., P:Ntfbn) is relatively open (141.2(2)°), and the P:N bond length is relatively short (1.514(2) Å). (((Trimethylsilyl)imino)(bis(dimethylamino))phosphorano)diazaphospho le gave, with Cp*TiCl3, [(η5-C5Me5)TiCl2(N:P(NMe2)2(2,5-dimethyl- 2H-1,2,3σ2diazaphosphol-4-yl))], which was also characterized structurally: monoclinic, P21 (Number 4), a = 11.9477(11) Å, b = 8.4757(6) Å, c = 12.7567(11) Å, $\beta =$ $108.824(8)^\circ$, V = 1222.7(2) Å3, Z = 2. Final indexes are R1 = 0.0630 and wR2 = 0.1593, and for all data, R1 = 0.0768, wR2 = 0.1973, and GOF = 1.081. The Ti-N-P angle of 161.0(5)° was large, and the P:N distance (1.592(6) Å) and the Ti-N distance (1.781(6) Å) were both slightly shorter than those in similar titanium complexes. The P-N single bond distances between the exo-phosphorus atom and the attached dimethylamino groups were also short (1.649 Å (average)). These short values suggest delocalized bonding character throughout the metal-liqand framework, possibly a consequence of addnl. conjugation through the diazaphosphole ring.

IT 229974-08-7P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(preparation and crystal structure of)

RN 229974-08-7 HCAPLUS

CN Phosphonimidic difluoride, N-(4-cyano-2,3,5,6-tetrafluorophenyl)-P-(2,5-dimethyl-2H-1,2,3-diazaphosphol-4-yl)- (CA INDEX NAME)

IT 229974-09-8P 229974-10-1P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

RN 229974-09-8 HCAPLUS

CN Phosphonimidic difluoride, P-(2,5-dimethyl-2H-1,2,3-diazaphosphol-4-yl)-N-[2,4,6-tris(1,1-dimethylethyl)phenyl]- (CA INDEX NAME)

November 13, 2009 10/540,558 27

Me Ne Me
$$P = P = F$$
 $P = F$ $P = F$

RN 229974-10-1 HCAPLUS

CN Phosphonimidic difluoride, P-(2,5-dimethyl-2H-1,2,3-diazaphosphol-4-yl)-N-(4-methylphenyl)- (CA INDEX NAME)

CC 29-7 (Organometallic and Organometalloidal Compounds) Section cross-reference(s): 75

IT 229974-08-7P 229974-17-8P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(preparation and crystal structure of)
IT 74582-13-1P 229974-07-6P 229974-09-8P

229974-10-1P 229974-11-2P 229974-12-3P 229974-13-4P 229974-14-5P 229974-15-6P 229974-16-7P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)
OS.CITING REF COUNT: 14 THERE ARE 14 CAPLUS RECORDS THAT CITE THIS

RECORD (14 CITINGS)

REFERENCE COUNT: 47 THERE ARE 47 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE

FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

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